Models of triple-stranded polynucleotides with optimised stereochemistry

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#### ABSTRACT

Detailed models are presented for the triple-stranded polynucleotide helices of  $poly(U) \cdot poly(A) \cdot poly(U)$  (two forms),  $poly(U) \cdot poly(A) \cdot poly(U)$ ,  $poly(U) \cdot poly(D) \cdot poly(D)$ ,  $poly(D) \cdot poly(D) \cdot poly(D) \cdot poly(D)$ ,  $poly(D) \cdot poly(D) \cdot poly(D) \cdot poly(D)$ . The models were generated using a computerized, linked-atom procedure which preserves standard bond lengths, bond angles and sugar ring conformations, constrains the helices to have the pitches and symmetries observed in X-ray diffraction experiments, and optimises the non-bonded interatomic contacts including hydrogen bonds. The possible biological significance of such complexes is discussed.

# INTRODUCTION

Triple-stranded complexes containing three coaxial helical chains connected by hydrogen-bonded bases can be formed from a variety of simple polynucleotides. In cases like poly d(T) poly  $d(A) \cdot poly d(T)$  two of the chains are hydrogen-bonded in the Watson-Crick manner: therefore local segments of a native DNA could participate in such complexes and it is of some interest to determine the effect of involving a third strand with such a duplex. Since, in the case mentioned, the second poly d(T) chain cannot be involved in Watson-Crick pairing, and in other cases like  $poly(I) \cdot poly(A) \cdot poly(I)$  none of basepairings can be of the standard kind, it is also of interest to determine the conformational consequences of having unusual base-base hydrogen bonds.

Although these complexes are, in many cases, susceptible to the same general methods of structure analysis as the more familiar double-stranded forms, several factors combine to make the extension of the technique problematic. Earlier determinations of nucleic acid structures<sup>3,4</sup> have used a linked-atom

description of the structure, that is, one in which bond lengths and angles are fixed and only the bond torsion angles are treated as explicit variables. This reduction of the number of parameters in the description allows the optimisation of the fit between the predicted X-ray diffraction of the model and the quite limited number of observed data. In triplestranded complexes, however, there no longer exists the simplification allowed by the symmetry between the strands of a double helix: the three strands each have different base hydrogen bonds and one cannot assume even that the sugar-phosphate backbones of the two strands joined by Watson-Crick base pairing have the same conformation. Coupled with this three-fold increase in the number of parameters to be determined is a scarcity of quantifiable X-ray diffraction data, caused by limited ordering in the samples studied.

To overcome these problems we must increasingly use stereochemical information obtained from simpler systems. One well-established and powerful constraint of this sort is the avoidance, where possible, of over-short interatomic distances between non-bonded atoms. This paper describes the use of a computer program to use this sort of information to predict probable molecular structures with standard bond lengths and angles for six triple-stranded polynucleotides. All but one of these have provided sufficient X-ray data for earlier analyses, thus affording a basis for assessment of the present method. Since nucleic acid systems of current biological interest are more complicated than the duplex structures that have received most attention heretofore it may be that the new strategy we will describe could form the basis for accurate modelling of those complex systems.

### METHOD AND PROCEDURE

The linked-atom least-squares procedure (LALS) has been described in detail earlier. 3,4,5,6,7 Briefly, it involves analytic minimisation of

$$\Phi = \sum_{j} k_{j} \delta_{j}^{2} + \sum_{h} \lambda_{h} G_{h}$$
 (1)

The first summation includes several classes of terms which are, together, to be minimised. The  $k_j$  are weights and the  $\delta_j$  are differences between model values and standard (i.e. known

or observed) values of various structural features. In the case of interatomic contacts (a contact being a separation *less* than the standard value), the standard values and weights are derived in advance by fitting a non-bonded van der Waals energy function. Where hydrogen bonds are known or suspected to exist, the standard value for calculating  $\delta_j$  is reduced to the known length of the hydrogen bond, whilst the original standard value is still used for determining whether the contact exists.

These terms, therefore, tend to drive apart atoms which are too close and to make hydrogen bonds close to standard lengths.

Other terms in the first summation include recognition of the known relationship that the angle  $(\gamma)$  between the helix axis and the normal to the plane of the bases in a polynucleotide bears to the projected height (h) of a residue on the helix axis.  $^{10}$ 

The parameters varied in order to minimise  $\Phi$  are, in the present case, primarily single bond torsion angles. There are also some geometrical parameters of no immediate physical significance needed to orient and position the structure and to allow sufficient flexibility to the hydrogen bonds between the strands.

The second summation in equation 1 is of linear Lagrange constraints ( $G_h$  = 0) on the parameters necessary to maintain the observed helical symmetry. The  $\lambda_i$  are undetermined multipliers.

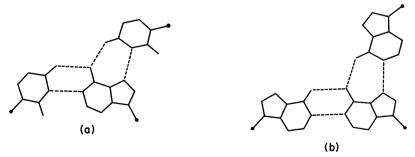
In each minimisation, consequently, 32 parameters were varied, upon which 27 constraints were applied of which 9 were linearly dependent on others, leaving 14 net degrees of freedom. The number of data used to determine these in this type of minimisation is not an obviously obtainable number, since one has to arbitrarily decide whether the lack of short contacts constitutes data, and if so, at what interatomic separation such data cease to be significant. The number of non-zero contributions to equation 1 was typically around 100 to 300, but it must be stressed that conventional considerations concerning the data-to-parameter ratio are not necessarily valid in this case, and a more appropriate value for 'number of

data', bearing in mind the desirability of a balanced distribution of  $\delta_{\it i}$ , might be several times greater.

Six complexes, listed in table 1, were analysed in the present study. Five of these have been investigated previously using other data and methods  $^{3,4,8,9}$  and have been shown to adopt similar hydrogen-bonding schemes, in which the third strand occupies what would be the major groove of an A-type polynucleotide duplex, and participates in Hoogsteen-type interactions with one of the two Watson-Crick linked strands (Figure 1). This, and the standard  $^{2}$ -endo sugar ring conformation observed in all polynucleotides of the  $^{4}$  genus, were assumed for the present study. Each of the six can be oriented sufficiently to show X-ray diffraction which yields the helix symmetry and the helix pitch, and hence the turn angle  $^{(t)}$  and projected residue height  $^{(h)}$ . (Table 1).

Table I: properties of the systems studied.

Complex	Abbreviation	Helix Symmetry	Pitch (nm)	Turn angle t (°)	Residue height h (nm)	Base skew γ(°)
Poly(U) · poly(A) · poly(U)	UAU-11	111	3.34	32.7	0.304	12.0
Poly(U) · poly(A) · poly(U)	UAU-12	121	3.65	30.0	0.304	12.0
Poly(U)· poly d(A)· poly(U)	U A DU	1111	3.34	32.7	0.304	12.0
Poly d(C). poly d(I). poly d(C)	dCdIdC	111	3.48	32.7	0.316	10.0
Poly d(T). poly d(A). poly d(T)	TDAbTb	121	3.91	30.0	0.326	8.5
Poly(I)· poly(A)· poly(I)	IAI	121	3.95	32.7	0.329	8.0



<u>Figure 1</u>: Triplex hydrogen-bonding schemes: (a) pyrimidine: <u>purine:pyrimidine</u>, (b) poly(I)·poly(A)·poly(I).

For each complex, six torsion angles in each strand were varied (Figure 2). Five of these define the sugar-phosphate backbone conformation, and the sixth is about the sugar-base bond.

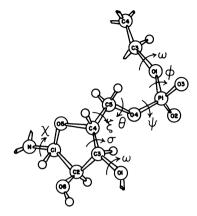


Figure 2: Atom and conformation angle nomenclature for a nucleotide residue. It should be noted that the above nomenclature we use differs from that in recent work, to correspond with that in the earlier studies considered in this paper. The conformation angles shown are defined as

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\omega = \theta [C4-C3-O1-P1] 

\phi = \theta [C3-O1-P1-O4] 

\psi = \theta [O1-P1-O4-C5] 

\theta = \theta [P1-O4-C5-C4] 

\xi = \theta [O4-C5-C4-C3] 

\chi = \theta [C2-C1-N-(C2pyr,C4pur)]
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### RESULTS AND DISCUSSION

In Table II are shown the values obtained for the varied torsion angles, and the differences between these and those obtained in the earlier studies. Although there are in some cases moderately large differences, it must be realised that the earlier structures were derived from quite limited numbers of data, supplemented (in most cases) by only statistical information concerning probable chain conformation angles.

Table III lists the hydrogen-bond lengths of the present models: none of these is exceptional.

Table IV shows the remaining short contacts in the present models. These are the interatomic distances between pairs of atoms not bonded (or hydrogen-bonded) to each other or to a common third atom, more than 0.02nm less than the sum of the

Table II:	_varied	confo	rmation	angles	(de	rees)	and di	fferen	es fr	om ear	lier s	tudies
Complex	ω	ф	ψ	θ	ξ	х	Δω	Δф	Δψ	Δθ	Δξ	ΔΧ
poly(U)·	-152	-74	- 55	173	47	83	13	-15	4	-16	3	- 3
poly(A)·	-156	- 69	-76	176	64	89	-2	9	16	- 4	-16	13
poly(U)	-149	-76	-63	174	53	83	-25	29	- 23	33	2	17
poly(U)·	-156	-75	- 28	171	24	87	16	-12	6	0	- 3	4
poly(A)·	-163	- 67	- 66	-179	53	91	II 5	-16	22	4	- 23	12
poly(U)	-149	-83	- 40	167	37	85	- 5	- 6	-10	12	3	10
poly(U)·	-152	-74	- 57	172	49	83	15	-18	2	-16	7	-1
poly(dA).	-160	- 67	- 71	178	58	92	-6	9	22	- 3	- 23	17
poly(U)	-149	-76	-62	174	53	84	-31	24	- 4	32	-13	-21
poly(dC)·	-155	-70	- 61	176	51	83	11					
poly(dI)·	-151	-72	-82	173	72	84	ll .					
poly(dC)	-153	-72	-65	178	54	84	ll .					
poly(dT)·	-156	-74	-46	174	39	86	12	-13	17	-12	- 8	- 3
poly(dA)·	-161	-68	-66	180	55	91	3	- 2	-16	1	15	-1
poly(dT)	-160	-77	-53	172	47	85	-5	-1-	- 5	Ō	4	-1
poly(I).	-152	-77	- 57	171	52	84	8	-11	16	-12	-11	4
poly(A)·	-154	- 75	-62	173	55	85	6	- 9	11	-10	- 8	5
poly(I)	-153	-75	-70	177	60	80	10	-14	-13	-11	15	- 3

The left hand part of this table lists the final values of the conformation angles in the present study. The right hand part lists the differences between these values and the earlier studies cited in the text. It should be noted that the earlier studies used a variety of different methods and data and that comparisons must be made with this in mind.

Table III: hydrogen-bond lengths (nm). For notation see Table IV.

•	UAU-11	UAU-12	UAAU	dCdIdC	dTdAdT	IAI
N1 A - N3 W'	0.28	0.28	0.28	0.28	0.28	-
N1 A - N1 W	-	-	-	-	-	0.28
N6 A - O4 W	0.30	0.29	0.30	-	0.31	-
N6 A - 06 W	-	-	-	-	-	0.30
06 A - N4 W	-	-	-	0.29	-	-
N7 A - N3 H	0.30	0.29	0.30	0.30	0.30	-
N7 A - N1 H	-	-	-	-	-	0.29
N6 A - O4 H	0.27	0.29	0.27	-	0.29	-
06 A - N4 H	-	-	-	0.27	-	-
N6 A - O6 H	-	-	-	-	-	0.29

van der Waals radii of the two atoms. The bracketed figures show these differences. Also shown are the final values  $(\Sigma)$  of the first summation of equation 1: these are on an arbitrary scale but show the relative stereochemical acceptability of the different structures.

Clearly the 12-fold  $poly(U) \cdot poly(A) \cdot poly(U)$  model is much less satisfactory than the others. This structure has always proved awkward to modelbuild using standard assumptions, and it may well be that some distortion of the sugar rings from the standard C3-endo shape is present in this system.

Of the other models a common feature of all except the all-

_	UAU-11	UAU-12	UADU	dCdIdC	dTdAdT	IAI
V-strand						
<del>23 cW</del> - 04 cW		0.27(0.03)				
25 cW - 01 cW		0.27(0.03)				
04 cW-HC6bW		-	0.23(0.02)	0.23(0.02)		
C2 s W - C5 s W*		0.28(0.04)		, ,		
C2 s W - HC5 s W*		0.19(0.08)			0.24(0.03)	
05 sW-C6 bW	0.27(0.03)	0.27(0.03)	0.27(0.03)	0.27(0.03)	0.27(0.03)	
D6 sW - C5 cW*		0.25(0.05)				
06 sW - HC5 cW*	0.22(0.03)	0.15(0.10)	0.23(0.02)			
1C2 s W - C6 b W*	0.25(0.02)	0.25(0.02)	0.24(0.03)		0.19(0.03)	
IC2 s W - HC6 b W*	0.19(0.03)	0.18(0.04)	0.19(0.03)	0.19(0.03)		
A-strand/H-strand						
02 cA - 02 bH	0.23(0.05)	0.23(0.05)	0.23(0.05)	0.26(0.02)	0.23(0.05)	
05 s H - 02 c A*		0.25(0.03)				
I-strand						
D4 с H - НС6 b Н	0.22(0.03)		0.22(0.03)	0.22(0.03)	0.23(0.02)	
1C5 s H - O6 s H*	` '	0.21(0.04)	• •	• •	, ,	
05 sH - C6 bH	0.27(0.03)	0.27(0.03)	0.27(0.03)	0.27(0.03)	0.27(0.03)	
HC2 s H - C6 b H		0.24(0.03)	-			
HC2 s H - HC6 b H		0.19(0.03)				
A-strand						
HC5 c A - 06 s A*		0.21(0.04)				

Table IV: Short contacts (nm) in the models. A indicates the central strand, W the Watson-Crick paired strand and H the Hoogsteen strand. The letters c, s and b signify chain, sugar

purine  $poly(I) \cdot poly(A) \cdot poly(I)$  is the contact between the phosphate O2 on the A strand and the base O2 on the H strand. Although this too may be an artefact of the standard assumptions, it has been shown that this interaction contributes to the anomalous infra-red spectrum from the base oxygen atoms of triple-stranded complexes,  $^{12}$  and it may therefore be at least partly a real feature of these molecules.

Generally, however, these models are stereochemically acceptable (with the exception of 12-fold poly(U) poly(A) poly(U)): much more so than those refined against X-ray data, whilst differing not greatly from the latter. We therefore conclude that this method is a valid and useful one for constructing such models.

Atomic coordinates of all six models are given in Table V, and a view perpendicular to the helix axis of a typical structure is shown in Figure 3.

We would further predict that models similar to those we have built for the pyrimidine:purine:pyrimidine systems would be possible and indeed probable for any such complexes. None of the DNA strands appears to have any serious interactions at the site of the ribose O6 atom, and it would seem therefore

Poly(U)·Poly(dA)·Poly(U) (11-Fold Helix)															
	x	· Y	z	R	•	x	Y	Z Z	R	•	x	Y	z	R	•
01 P 02 03 04	.1294 .1392 .1545 .0259 .2754	9204 8654 9786 7740 7817	.0701 .2200 .3140 .2469 .2141	.9295 .8765 .9908 .7744 .8288	-82.00 -80.86 -81.03 -88.08 -70.59	.1579 .1818 .0635 .2223 .3057	.9458 .9011 .8268 1.0185 .8012	0327 1845 2335 2650 1689	.9589 .9193 .8293 1.0424 .8575	80.52 78.60 85.61 77.69 69.11	8224 7538 8477 7002 6328	.4595 .4445 .4894 .3075 .5485	2509 3946 4998 4102 3828	.9421 .8751 .9789 .7647 .8374	150.81 149.47 150.00 156.29 139.08
C1 C2 C3 C4 C5 O5	.5218 .5937 .5273 .5079 .3965 .4764 .7302	5293 5300 6487 7469 8481 6614 5572	.0673 .2018 .2714 .1559 .1732 .0421	.7433 .7959 .8360 .9032 .9362 .8151	-45.41 -41.76 -50.89 -55.78 -64.95 -54.23 -37.35	.5251 .6052 .5540 .5358 .4351 .4899	.5180 .5236 .6535 .7427 .8550 .6511	0524 1820 2442 1214 1356 0178	.7376 .8003 .8567 .9159 .9593 .8148	44.61 40.87 49.71 54.19 63.03 53.04	3318 2982 4244 5341 6623 4731 2850	.7201 .7794 .7466 .7653 .6881 .7179 .9193	2277 3641 4438 3390 3631 2154 3462	.7929 .8345 .8587 .9332 .9550 .8598	114.74 110.94 119.61 124.91 133.91 123.39 107.23
N1 C2 N3 C4 C5 C6 N7 C8 N9 N6 O2	.4038 .4281 .3163 .1856 .1702 .2771 	4385 3055 2244 2644 4058 469   2608 1824	.0630 .0385 .0352 .0540 .0790 .0827	.5961 .5259 .3879 .3230 .4400 .5602       	-47.36 -35.51 -35.36 -54.94 -67.25 -60.35   -25.78 -62.84	.3118 .4356 .4830 .3860 .2551 .2155 .1859 .2756 .3995	.0548 .1003 .2217 .3052 .2745 .1417 .3869 .4805 .4385	.0077 .0260 .0082 0344 0579 0356 1009 1020 0630	.3165 .4470 .5314 .4921 .3747 .2579 .4292 .5539 .5932	9.97 12.96 24.65 38.33 47.10 33.32 64.34 60.16 47.66 47.21	2818 1512 1084 1836 3194 3636 	. \$810 . \$664 . 4361 . 3221 . 3476 . 4733 	2089 1688 1524 1723 2142 2309     -1487 1544	.6457 .5862 .4494 .3708 .4721 .5969 	115.87 104.95 103.96 119.69 132.58 127.53
	Poly(dT) -Poly(dA) -Poly(dT) (12-Fold Helix)														
	x	Y	z	R	•	x	Y	Z	R	•	x	Y	Z	R	•
01 P 02 03 04	.1823 .1814 .1905 .0662 .3174	9346 8757 9865 7843 7914	.0544 .2031 .3008 .2197 .2047	.9522 .8943 1.0047 .7870 .8527	-78.96 -78.30 -79.07 -85.18 -68.15	.2770 .2883 .1618 .3333 .4046	.9669 .9182 .8530 1.0307 .8088	0185 1705 2112 2555 1601	1.0058 .9624 .8683 1.0833 .9043	74.01 72.57 79.26 72.08 63.43	7287 6757 7656 6533 5349	.6505 .6176 .6787 .4719 .6937	1657 3130 4134 3265 3137	.9768 .9154 1.0232 .8060 .8760	138.25 137.57 138.44 144.16 127.64
C1 C2 C3 C4 C5	.5722 .6300 .5570 .5494 .4369 .5297	5278 5354 6571 7494 8509 6582	.0740 .2149 .2714 .1498 .1507 .0378	.7784 .8268 .8614 .9292 .9565 .8449	-42.69 -40.36 -49.72 -53.75 -62.82 -51.18	.6079 .6803 .6350 .6307 .5376	.5064 .5105 .6456 .7319 .8514	0572 1913 2465 1204 1254 0168	.7912 .8506 .9056 .9662 1.0070 .8671	39.80 36.88 45.47 49.25 57.73 47.62	1895 1573 2949 3867 5306 3258	.7881 .8411 .8387 .8792 .8328 .8170	1757 3150 3813 2659 2766 1490	.8106 .8557 .8891 .9605 .9875 .8796	103.52 100.59 109.37 113.74 122.50 111.74
N1 C2 N3 C4 C5 C6 N7 C8 N9 N6 O2 O4 Me	. 4551 . 4816 . 3708 . 2388 . 2211 . 3272 	4363 3032 2216 2611 4026 4843 	.0622 .0408 .0305 .0394 .0619 .0724 	.6305 .5692 .4319 .3539 .4594 .5845 	-43.79 -32.19 -30.87 -47.55 -61.23 -55.96 	. 3666 . 4944 . 5490 . 4555 . 3214 . 2740 . 2575 . 3535 . 4765 . 1449	.0562 .0921 .2108 .3029 .2825 .1517 .4015 .4887 .4364	0040 .0071 0081 0393 0543 0356 0864 0897 0622 0473	. 3709 . 5029 . 5881 . 5470 . 4279 . 3132 . 4770 . 6031 . 6461 . 1865	8.72 10.56 21.01 33.63 41.32 28.97 57.33 54.12 42.48 38.98	1701 0434 0293 1289 2583 2749    0509 1041 3723	.6410 .5973 .4605 .3662 .4217 .5544   .6730 .2461	1616 1313 1191 1343 1661 1785 	. 6632 . 5989 . 4614 . 3882 . 4945 . 6188 	104.86 94.16 93.64 109.39 121.49 116.37 
						P	oly(I)·F	oly(A).F							
	x	Y	z	· R	•	x	Y	z	R	•	x	Y	Z	R	φ
01 P 02 03 04	.0535 .0604 .0602 0453 .2038	9875 9378 -1.0547 8371 8671	.0274 .1794 .2702 .2036 .1831	.9889 .9398 1.0564 .8383 .8908	-86.90 -86.32 -86.73 -93.10 -76.77	.0721 .0825 0267 .0910 .2228	.9976 .9498 .8544 1.0679 .8728	0206 1729 2026 2616 1723	1.0002 .9533 .8548 1.0718 .9008	85.87 85.03 91.79 85.13 75.68	9938 9282 -1.0341 8396 8400	.2225 .2292 .2481 .1124 .3622	1615 3073 4089 3274 2962	1.0184 .9561 1.0635 .8471 .9148	167.38 166.13 166.51 172.38 156.67
C1 C2 C3 C4 C5 O5	.4751 .5451 .4675 .4406 .3203 .4183	6330 6462 7613 8521 9435 7592 6844	.0558 .1907 .2545 .1344 .1464 .0244	.7915 .8454 .8934 .9592 .9964 .8668	-53.11 -49.85 -58.45 -62.66 -71.25 -61.14 -45.23	.4871 .5590 .4855 .4599 .3425 .4337 .6936	.6297 .6430 .7613 .8510 .9461 .7570	0530 1868 2496 1284 1402 0201 1581	.7962 .8520 .9030 .9673 1.0062 .8724 .9691	52.28 49.00 57.47 61.61 70.10 60.19 44.30	5918 5845 7004 8047 9066 7256 6109	.6149 .6814 .6139 .6003 .4896 .5723	1706 3077 3808 2698 2881 1506 2880	.8534 .8977 .9313 1.0039 1.0304 .9242 1.0219	133.90 130.62 138.77 143.28 151.63 141.74 126.71
N1 C2 N3 C4 C5 C6 N7 C8 N9 N6	. 3603 . 4811 . 4996 . 3838 . 2601 . 2380 . 1637 . 2333 . 3658	1444 2065 3348 4006 3436 2075 4396 5532 5532	.0061 .0010 .0153 .0361 .0422 .0273 .0647 .0717	.3881 .5235 .6014 .5547 .4310 .3158 .4691 .6004 .6455	-21.84 -23.23 -33.83 -46.23 -52.88 -41.08 -69.57 -67.13 -55.47	. 3566 . 4733 . 4989 . 3857 . 2591 . 2434 . 1685 . 2421 . 3749 . 1251	.1356 .1993 .3275 .3966 .3466 .2081 .4486 .5553 .518	0034 .0036 0104 0349 0451 0284 0711 0757 0547	. 3815 . 5136 . 5968 . 5532 . 4327 . 3203 . 4792 . 6058 . 65506	20.82 22.84 33.29 45.80 53.23 40.52 69.41 66.45 54.82 49.33	1210 1698 2947 3717 3283 1961 4330 5380 5380	.4532 .5799 .6108 .5013 .3722 .3371 .2850 .3655 .4959	0889 0952 1171 1332 1279 1048 1486 1659 1570	. 4691 .6043 .6781 .6240 .4963 .3899 .5184 .6504	104.95 106.32 115.76 126.56 131.41 120.18 146.64 145.81 135.41

dues may be generated by applying to the listed coordinates a translation of h along and a rotation of t about the helix axis, z. The accuracy of these coordinates is not necessarily that

	Poly(U) · Poly(A) · Poly(U) (11 · Fold Helix)														
	x	, <b>Y</b>	Z	R	•	x	Y	z	R	•	x	Y	Z	R	•
01 P 02 03 04	.0844 .0962 .1035 0114 .2374	9186 8610 9731 7622 7862	.0808 .2297 .3261 .2535 .2236	.9225 .8664 .9785 .7623 .8212	-84.75 -83.63 -83.93 -90.86 -73.20	.2544 .2728 .1482 .3224 .3880	.9419 .8937 .8292 1.0062 .7836	0427 1941 2413 2764 1784	.9757 .9344 .8423 1.0566 .8744	74.89 73.02 79.87 72.23 63.66	7653 6972 7847 6584 5658	.5500 .5269 .5809 .3848 .6174	2576 4004 5069 4146 3879	.9425 .8739 .9763 .7626 .8375	144.30 142.92 143.49 149.69 132.50
C1 C2 C3 C4 C5 O5	.5009 .5696 .4944 .4714 .3536 .4479 .7047	5503 5541 6677 7656 8596 6795 5900	.0747 .2108 .2799 .1648 .1805 .0496	.7441 .7947 .8308 .8991 .9295 .8139	-47.69 -44.21 -53.48 -58.38 -67.64 -56.61 -39.94	.5815 .6645 .6264 .6137 .5237 .5576 .8006	.4879 .4802 .6108 .7072 .8271 .6250	0568 1846 2539 1358 1578 0293 1454	.7591 .8198 .8749 .9364 .9789 .8376	40.00 35.85 44.28 49.05 57.66 48.26 31.04	2493 2087 3372 4450 5805 3902 1809	.7578 .8116 .7915 .8230 .7596 .7707	2324 3692 4494 3457 3699 2211 3525	.7977 .8380 .8604 .9355 .9560 .8639	108.21 104.42 113.07 118.40 127.39 116.85 100.79
N1 C2 N3 C4 C5 C6 N7 C8 N9 N6 O2	.3888 .4219 .3155 .1822 .1574 .2590         	4524 3214 2336 2652 4050 4926 	.0669 .0418 .0353 .0514 .0773 .0841	.5965 .5304 .3926 .3217 .4346 .5566  .6073 .2018	-49.32 -37.30 -36.51 -55.51 -68.76 -62.26       -27.87 -61.71	.3300 .4574 .5147 .4250 .2918 .2412 .2321 .3295 .4496	.0443 .0791 .1960 .2876 .2682 .1391 .3861 .4718 .4193 .1059	.0035 .0212 .0034 0385 0613 0390 1036 1051 0669	. 3329 . 4641 . 5508 . 5131 . 3963 . 2785 . 4505 . 5754 . 6148 . 1545	7.64 9.81 20.85 34.09 42.59 29.97 58.99 55.07 43.01	2145 0858 0573 1446 2773 3078    	.6143 .5862 .4523 .3469 .3864 .5159   .6725 .2309	2118 1727 1546 1719 2131 2314     1548 1527	.6507 .5925 .4559 .3758 .4756 .6007 	109.25 98.33 97.22 112.63 125.67 120.82
						P		oly(A)·P							
	х	Y	2	R	•	x	(12-F	old Heli: Z	k) R	φ	x	Y	Z	R	¢
01 P 02 03 04	.1493 .1504 .1447 .0453 .2946	9219 8593 9677 7556 7901	.0955 .2427 .3432 .2534 .2465	.9340 .8723 .9785 .7570 .8432	-80.80 -80.07 -81.50 -86.57 -69.55	.3260 .3426 .2155 .3967 .4532	.9729 .9208 .8604 1.0295 .8066	0532 2036 2495 2881 1854	1.0261 .9824 .8870 1.1033 .9252	71.48 69.59 75.94 68.93 60.67	6637 6113 7002 5919 4692	.7118 .6698 .7270 .5232 .7430	2314 3767 4803 3821 3815	.9732 .9069 1.0094 .7900 .8788	132.99 132.39 133.92 138.52 122.27
C1 C2 C3 C4 C5 O5	.5620 .6144 .5305 .5201 .4002 .5110	5330 5436 6588 7514 8442 6602 5805	.0904 .2332 .2882 .1670 .1649 .0537	.7745 .8203 .8459 .9139 .9343 .8349	-43.48 -41.50 -51.15 -55.31 -64.63 -52.26 -37.70	.6412 .7158 .6769 .6741 .5861 .6227	.4979 .4963 .6319 .7210 .8440 .6336	0683 2013 2600 1357 1445 0310 1709	.8118 .8710 .9260 .9870 1.0275 .8884 .9860	37.83 34.74 43.03 46.92 55.22 45.50 29.97	1235 0829 2179 3107 4570 2584 0287	.8110 .8574 .8622 .9124 .8756 .8499 .9875	2117 3512 4225 3118 3265 1910 3371	.8203 .8614 .8893 .9639 .9877 .8883	98.66 95.52 104.18 108.81 117.56 106.91 91.67
N1 C2 N3 C4 C5 C6 N7 C8 N9 N6 O2	.4526 .4898 .3858 .2511 .2220 .3212 	4331 3030 2133 2423 3815 4708 	.0740 .0504 .0357 .0425 .0676 .0823 	.6265 .5759 .4409 .3490 .4414 .5700	-43.74 -31.74 -28.93 -43.98 -59.81 -55.70   -23.82 -42.57	.3807 .5096 .5693 .4803 .3458 .2927 .2874 .3870 .5072 .1626	.0596 .0907 .2065 .3014 .2860 .1578 .4064 .4895 .4331 .1283	0072 .0055 0120 0477 0649 0435 1017 1056 0740	.3854 .5176 .6056 .5670 .4487 .3325 .4978 .6240 .6669 .2071	8.90 10.09 19.94 32.11 39.60 28.32 54.73 51.68 40.50 38.28	1148 .0078 .0120 0936 2182 2251   .1069 0777	.6634 .6120 .4749 .3874 .4508 .5838	1927 1582 1415 1562 1925 2094 	. 6733 . 6120 . 4751 . 3985 . 5008 . 6257 	99.82 89.27 88.55 103.59 115.83 111.08
						Pol		ly(dI)·F							
	x	Y	z	R	φ •	x	(11-F0 Y	ld Helix Z	r R	ф	x	Y	Z	ĸ	ф
01 P 02 03 04	.1335 .1455 .1660 .0309 .2798	9155 8592 9715 7703 7725	.0484 .1977 .2920 .2265 .1879	.9252 .8714 .9856 .7709 .8216	-81.70 -80.39 -80.30 -87.70 -70.09	.2217 .2376 .1142 .2809 .3572	.9605 .9152 .8471 1.0309 .8094	0174 1702 2151 2520 1594	.9857 .9455 .8548 1.0685 .8847	77.00 75.45 82.32 74.76 66.18	8055 7364 8280 6872 6123	.4692 .4554 .5058 .3171 .5551	2144 3581 4628 3765 3428	.9322 .8659 .9702 .7568 .8265	149.78 148.27 148.58 155.23 137.81
C1 C2 C3 C4 C5 O5	.5301 .5985 .5307 .5147 .4032 .4860 .7358	5169 5202 6405 7365 8382 6487 5465	.0578 .1942 .2596 .1416 .1538 .0289	.7404 .7930 .8318 .8985 .9302 .8105 .9166	-44.28 -41.00 -50.36 -55.05 -64.31 -53.16 -36.60	.5631 .6500 .6080 .5876 .4928 .5321	.5249 .5165 .6430 .7425 .8573 .6615	0407 1660 2408 1267 1556 0191 1228	.7698 .8302 .8850 .9469 .9889 .8490	42.99 38.47 46.61 51.64 60.11 51.19 33.84	3004 2716 4020 5065 6374 4411 2545	.7219 .7781 .7472 .7702 .6958 .7233	2057 3445 4176 3083 3254 1869 3296	.7819 .8242 .8485 .9218 .9437 .8472	112.59 109.24 118.28 123.33 132.49 121.38 105.49
N1 C2 N3 C4 C5 C6 N7 C8 N9 O6 O2 N4	.4120 .4366 .3312 .2060 .1784 .2853  .5535	4265 2913 2058 2508 3894 4733 	.0522 .0303 .0244 .0397 .0626 .0679	.5931 .5248 .3899 .3245 .4284 .5526 	-45.99 -33.71 -31.85 -50.59 -65.38 -58.92      -24.60 -58.25	.3379 .4676 .5157 .4207 .2890 .2421 .2200 .3132 .4346 .1288	.0756 .1101 .2308 .3191 .2901 .1555 .4034 .4989 .4499	.0021 .0216 .0077 -0292 -0501 -0331 -0867 -0867 -0522 -0477	.3462 .4804 .5650 .5280 .4095 .2878 .4595 .5890 .6255 .1701	12.61 13.25 24.12 37.18 45.11 32.71 61.40 57.88 45.99 40.80	2530 1199 0729 1538 2915 3361    0480 1079	. 5818 . 5645 . 4383 . 3326 . 3484 . 4755 	1865 1507 1326 1489 1858 2036 	.6345 .5771 .4444 .3665 .4543 .5823 	113.50 101.99 99.44 114.81 129.91 125.25 

implied by the precision to which they are quoted. Because of the method used there is very strong correlation between the errors in coordinates of atoms closely joined by chemical bonds, and this correlation falls off as more variable structural features intervene between the atoms considered.

For this reason and others concerned with the statistical properties of the system, no error estimates are presented here. A fuller discussion is in preparation (ref. 19).



Figure 3: View perpendicular to the helix axis of a typical triplex structure, the poly(dT) poly(dA) poly(dT) of Selsing and Arnott (ref. 4). Notice how the third strand (shown shaded) fits in the major groove of a quite conventional A-DNA type structure.

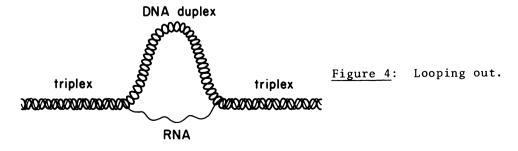
that this conformational class would be available to DNAs, RNAs, and DNA-RNA hybrids: for example the triple-stranded complex of poly d(A-G) poly d(C-T) and poly(C-U).  $^{13}$ 

No biological role for such triple-stranded complexes has yet been demonstrated, but we would be surprised if analogous structures were not exploited by living systems. It has indeed been suggested that DNA-RNA triplexes of this sort may serve to maintain chromosomes in folded conformations because folded E. coli chromosomes relax after treatment with ribonuclease speci-

fic for single-stranded RNA.  $^{14,15}$  As shown in Figure 4, poly(Py) single-stranded RNA could form triplexes with two separated poly(Pu)·poly(Py) tracts in the chromosome; the bounded duplex would then loop-out. The triplex regions of this structure would be required to adopt A-type conformations; presumably the looped-out duplex would retain the B-DNA geometry, with kinks occurring at the junction between the A and B regions. Of course this scheme requires the presence of sizable oligo(Pu)·oligo(Py) tracts in the chromosome; it is noteworthy, therefore, that long sequences ( $\sim$ 750 base pairs on the average) of this type are found in the Drosophila genome.  $^{16,17}$ 

The discovery 18 that small RNA oligonucleotides are implicated in the initiation of DNA replication in the bacteriophage λ invites speculation that three-stranded structures are involved in this process. The geometry of three-stranded initiator regions, however, would preclude primer roles for these oligonucleotides since the 3'-hydroxyl group of the RNA, located in the groove of the DNA duplex (Figure 1), is not positioned correctly for chain elongation, since the oligo RNA would have the same 3'-5' polarity as the poly(purine) nucleotide stretch to which it would be attached. Rather we envisage the oligopyrimidine nucleotide tracts of such RNAs acting as inducers of an A conformation in a DNA that would normally have a B secondary structure. Triplex formation would require the local conformation to be A. The region of A conformation might well extend beyond the triplex region to a nearby site and enhance polymerase binding.

In conclusion, it seems to us that this method is a valid one for the postulation and evaluation of the structures of these interesting polynucleotides, a role played at present by



no more direct method. The paucity of X-ray data arises in part from the computational difficulty of extracting structural information from 'continuous' diffraction, but more fundamentally from the inherent disorder present in synthetic and, to an even greater extent, biological polynucleotide materials. It is of some importance then that such methods as the present one be developed.

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